1. Chemical Product and Company Identification

<table>
<thead>
<tr>
<th>Product Name:</th>
<th>Sulfur Hexafluoride (MSDS No. P-4657-C)</th>
<th>Trade Name:</th>
<th>Sulfur Hexafluoride, Sulfur Fluoride</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chemical Name:</td>
<td>Sulfur Hexafluoride</td>
<td>Synonyms:</td>
<td>Sulfur Fluoride</td>
</tr>
<tr>
<td>Formula:</td>
<td>SF₆</td>
<td>Chemical</td>
<td>Nonmetal Halide</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Family:</td>
<td></td>
</tr>
<tr>
<td>Telephone:</td>
<td>Emergencies:</td>
<td>Company Name:</td>
<td>Praxair, Inc.</td>
</tr>
<tr>
<td></td>
<td>1-800-645-4633*</td>
<td></td>
<td>39 Old Ridgebury Road</td>
</tr>
<tr>
<td></td>
<td>CHEMTREC</td>
<td></td>
<td>Danbury CT 06810-5113</td>
</tr>
<tr>
<td></td>
<td>1-800-424-9300*</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Routine:</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>1-800-PRAXAIR</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

*Call emergency numbers 24 hours a day only for spills, leaks, fire, exposure, or accidents involving this product. For routine information contact your supplier, Praxair sales representative, or call 1-800-PRAXAIR (1-800-772-9247).

2. Composition / Information on Ingredients

For custom mixtures of this product request a Material Safety Data Sheet for each component. See Section 16 for important information about mixtures.

<table>
<thead>
<tr>
<th>INGREDIENT NAME</th>
<th>CAS NUMBER</th>
<th>PERCENTAGE</th>
<th>OSHA PEL</th>
<th>ACGIH TLV</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sulfur Hexafluoride</td>
<td>2551-62-4</td>
<td>&gt;99%*</td>
<td>1000 ppm</td>
<td>1000 ppm</td>
</tr>
</tbody>
</table>

*The symbol ">" means "greater than."

3. Hazards Identification

**EMERGENCY OVERVIEW**

CAUTION! Liquid and gas under pressure. Can cause rapid suffocation. May cause frostbite. May cause dizziness and drowsiness. Self-contained breathing apparatus may be required by rescue workers. Odor: None

**THRESHOLD LIMIT VALUE:** 1000 ppm (ACGIH 1997). TLV-TWAs should be used as a guide in the control of health hazards and not as fine lines between safe and dangerous concentrations.
EFFECTS OF A SINGLE (ACUTE) OVEREXPOSURE:

INHALATION—Asphyxiant. High concentrations can cause headache, drowsiness, dizziness, excitation, excess salivation, vomiting, and unconsciousness. Lack of oxygen can kill.

SKIN CONTACT—No harm expected from vapor. Liquid may cause frostbite.

SWALLOWING—An unlikely route of exposure, but frostbite of the lips and mouth may result from contact with the liquid. This product is a gas at normal temperature and pressure.

EYE CONTACT—No harm expected from vapor. Liquid may cause frostbite.

EFFECTS OF REPEATED (CHRONIC) OVEREXPOSURE: No harm expected.

OTHER EFFECTS OF OVEREXPOSURE: Decomposition products generated at high temperatures may be irritating.

MEDICAL CONDITIONS AGGRAVATED BY OVEREXPOSURE: The toxicology and the physical and chemical properties of Sulfur Hexafluoride suggest that overexposure is unlikely to aggravate existing medical conditions.

SIGNIFICANT LABORATORY DATA WITH POSSIBLE RELEVANCE TO HUMAN HEALTH HAZARD EVALUATION: None known.

CARCINOGENICITY: Sulfur Hexafluoride is not listed by NTP, OSHA, or IARC.

4. First Aid Measures

INHALATION: Remove to fresh air. Give artificial respiration if not breathing. If breathing is difficult, qualified personnel may give oxygen. Call a physician.

SKIN CONTACT: For exposure to liquid, immediately warm frostbite area with warm water not to exceed 105°F (41°C). In case of massive exposure, remove clothing while showering with warm water. Call a physician.

SWALLOWING: This product is a gas at normal temperature and pressure.

EYE CONTACT: For exposure to liquid, immediately flush eyes thoroughly with warm water for at least 15 minutes. Hold the eyelids open and away from the eyeballs to ensure that all surfaces are flushed thoroughly. See a physician, preferably an ophthalmologist, immediately.

NOTES TO PHYSICIAN: There is no specific antidote. This product is inert. Treatment of overexposure should be directed at the control of symptoms and the clinical condition.

5. Fire Fighting Measures

<table>
<thead>
<tr>
<th>FLASH POINT (test method)</th>
<th>Not applicable</th>
<th>AUTOIGNITION TEMPERATURE</th>
<th>Not applicable</th>
</tr>
</thead>
<tbody>
<tr>
<td>FLAMMABLE LIMITS IN AIR, % by volume</td>
<td>LOWER</td>
<td>Not applicable</td>
<td>UPPER</td>
</tr>
</tbody>
</table>

EXTINGUISHING MEDIA: Sulfur Hexafluoride cannot catch fire. Use media appropriate for surrounding fire.
SPECIAL FIRE FIGHTING PROCEDURES:
CAUTION! Liquid and gas under pressure. Evacuate all personnel from danger area. Immediately deluge cylinders with water from maximum distance until cool, then move them away from fire area if without risk. Self-contained breathing apparatus may be required by rescue workers. On-site fire brigades must comply with OSHA 29 CFR 1910.156.

UNUSUAL FIRE AND EXPLOSION HAZARDS: Sulfur Hexafluoride cannot catch fire. Heat of fire can build pressure in cylinder and cause it to rupture. No part of cylinder should be subjected to a temperature higher than 125°F (52°C). Sulfur Hexafluoride cylinders are equipped with a pressure relief device. (Exceptions may exist where authorized by DOT.)

HAZARDOUS COMBUSTION PRODUCTS: See section 10.

6. Accidental Release Measures

STEPS TO BE TAKEN IF MATERIAL IS RELEASED OR SPILLED:
CAUTION! Liquid and gas under pressure. Immediately evacuate all personnel from danger area. Use self-contained breathing apparatus where needed. Shut off flow if without risk. Ventilate area or move cylinder to a well-ventilated area. Before entering area, especially confined areas, check for sufficient oxygen with an appropriate device. Prevent runoff from contaminating surrounding environment.

WASTE DISPOSAL METHOD: Prevent waste from contaminating the surrounding environment. Keep personnel away. Discard any product, residue, disposable container or liner in an environmentally acceptable manner, in full compliance with federal, state, and local regulations. If necessary, call your local supplier for assistance.

7. Handling and Storage

PRECAUTIONS TO BE TAKEN IN STORAGE: Store and use with adequate ventilation. Firmly secure cylinders upright to keep them from falling or being knocked over. Screw valve protection cap firmly in place by hand. Store only where temperature will not exceed 125°F (52°C). Store full and empty cylinders separately. Use a first-in, first-out inventory system to prevent storing full cylinders for long periods.

PRECAUTIONS TO BE TAKEN IN HANDLING: Protect cylinders from damage. Use a suitable hand truck to move cylinders; do not drag, roll, slide, or drop. Never attempt to lift a cylinder by its cap; the cap is intended solely to protect the valve. Never insert an object (e.g., wrench, screwdriver, pry bar) into cap openings; doing so may damage the valve and cause a leak. Use an adjustable strap wrench to remove over-tight or rusted caps. Open valve slowly. If valve is hard to open, discontinue use and contact your supplier. For other precautions in using Sulfur Hexafluoride, see section 16.

8. Exposure Controls/Personal Protection

VENTILATION/ENGINEERING CONTROLS:

LOCAL EXHAUST—Use a local exhaust system, if necessary, to control the concentration of this product in the worker's breathing zone.

MECHANICAL (general)—General exhaust ventilation may be acceptable if it can maintain an adequate supply of air.

SPECIAL—None

OTHER—None
RESPIRATORY PROTECTION: Use air-supplied respirators where the local and/or general exhaust ventilation are insufficient to keep worker's exposure below the TLV. Use a full-face, self-contained breathing apparatus in a positive pressure demand mode in confined spaces or oxygen deficient atmospheres. Respiratory protection use must conform with OSHA rules as specified in 29 CFR 1910.134.

SKIN PROTECTION: Wear work gloves when handling cylinders and to prevent exposure to liquid.

EYE PROTECTION: Wear safety glasses when handling cylinders. Select in accordance with OSHA 29 CFR 1910.133.


### 9. Physical and Chemical Properties

<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>MOLECULAR WEIGHT</td>
<td>146.05</td>
</tr>
<tr>
<td>SPECIFIC GRAVITY (air=1):</td>
<td>68°F (20°C) and 1 atm: 5.11</td>
</tr>
<tr>
<td>GAS DENSITY:</td>
<td>At 70°F (21.1°C) and 1 atm: 0.385 lbs/ft³ (6.17 kg/m³)</td>
</tr>
<tr>
<td>PERCENT VOLATILES BY VOLUME:</td>
<td>100</td>
</tr>
<tr>
<td>BOILING POINT (1 atm):</td>
<td>Sublimes @ -83°F (-63.9°C)</td>
</tr>
<tr>
<td>MELTING POINT @ 325 psia (2241 kPa abs):</td>
<td>-59.4°F (-50.8°C)</td>
</tr>
<tr>
<td>APPEARANCE, ODOR, AND STATE:</td>
<td>Colorless gas at normal temperature and pressure.</td>
</tr>
<tr>
<td>EXPANSION RATIO:</td>
<td>Not applicable</td>
</tr>
<tr>
<td>SOLUBILITY IN WATER:</td>
<td>Negligible</td>
</tr>
<tr>
<td>VAPOR PRESSURE:</td>
<td>AT 70°F (21.1°C): 312.7 psia (2156 kPa abs)</td>
</tr>
<tr>
<td>EVAPORATION RATE (Butyl Acetate=1):</td>
<td>High</td>
</tr>
<tr>
<td>pH:</td>
<td>Not applicable</td>
</tr>
</tbody>
</table>

### 10. Stability and Reactivity

<table>
<thead>
<tr>
<th>Reactivity</th>
<th>Unstable</th>
<th>Stable</th>
<th>X</th>
</tr>
</thead>
<tbody>
<tr>
<td>STABILITY:</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>INCOMPATIBILITY (materials to avoid): Explodes violently in contact with disilane.</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>HAZARDOUS DECOMPOSITION PRODUCTS: Thermal decomposition may produce toxic fumes of fluorides and sulfur oxides.</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>HAZARDOUS POLYMERIZATION: May Occur</td>
<td>Will Not Occur</td>
<td>X</td>
<td></td>
</tr>
<tr>
<td>CONDITIONS TO AVOID: Temperatures in excess of 1472°F (800°C).</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

### 11. Toxicological Information

Sulfur Hexafluoride is a simple asphyxiant.
12. Ecological Information

No adverse ecological effects expected. Sulfur Hexafluoride does not contain any Class I or Class II ozone-depleting chemicals. Sulfur Hexafluoride is not listed as a marine pollutant by DOT.

13. Disposal Considerations

WASTE DISPOSAL METHOD: Do not attempt to dispose of residual or unused quantities. Return cylinder to supplier.

14. Transport Information

DOT/IMO SHIPPING NAME: Sulfur hexafluoride
HAZARD CLASS: 2.2
IDENTIFICATION NUMBER: UN 1080
PRODUCT RQ: None
SHIPPING LABEL(s): NONFLAMMABLE GAS
PLACARD (When required): NONFLAMMABLE GAS

SPECIAL SHIPPING INFORMATION: Cylinders should be transported in a secure position, in a well-ventilated vehicle. Cylinders transported in an enclosed, nonventilated compartment of a vehicle can present serious safety hazards.

Shipment of compressed gas cylinders that have been filled without the owner's consent is a violation of federal law [49 CFR 173.301(b)].

15. Regulatory Information

The following selected regulatory requirements may apply to this product. Not all such requirements are identified. Users of this product are solely responsible for compliance with all applicable federal, state, and local regulations.

U.S. FEDERAL REGULATIONS:

EPA (Environmental Protection Agency)
  Reportable Quantity (RQ): None
SARA: Superfund Amendment and Reauthorization Act:
  • SECTIONS 302/304: Require emergency planning based on Threshold Planning Quantity (TPQ) and release reporting based on Reportable Quantities (RQ) of extremely hazardous substances (40 CFR Part 355):
    Threshold Planning Quantity (TPQ): None.
- **SECTIONS 311/312:** Require submission of Material Safety Data Sheets (MSDSs) and chemical inventory reporting with identification of EPA hazard categories. The hazard categories for this products are as follows:

  IMMEDIATE: Yes
  DELAYED: No
  PRESSURE: Yes
  REACTIVITY: No
  FIRE: No

- **SECTION 313:** Requires submission of annual reports of release of toxic chemicals that appear in 40 CFR Part 372.

  Sulfur Hexafluoride does not require reporting under Section 313.

- **40 CFR 68:** Risk Management Program for Chemical Accidental Release Prevention: Requires development and implementation of risk management programs at facilities that manufacture, use, store, or otherwise handle regulated substances in quantities that exceed specified thresholds.

  Sulfur Hexafluoride is not listed.

- **TSCA:** Toxic Substances Control Act: Sulfur Hexafluoride is listed on the TSCA inventory.

- **OSHA (OCCUPATIONAL SAFETY AND HEALTH ADMINISTRATION):**

  - **29 CFR 1910.119:** Process Safety Management of Highly Hazardous Chemicals: Requires facilities to develop a process safety management program based on Threshold Quantities (TQ) of highly hazardous chemicals.

    Sulfur Hexafluoride is not listed in Appendix A as a highly hazardous chemical.

- **STATE REGULATIONS:**

  **CALIFORNIA:** This product is not listed by California under the Safe Drinking Water Toxic Enforcement Act of 1986 (Proposition 65).

  **PENNSYLVANIA:** This product is subject to the Pennsylvania Worker and Community Right-To-Know Act (35 P.S. Sections 7301-7320).

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### 16. Other Information

Be sure to read and understand all labels and instructions supplied with all containers of this product.

**SPECIAL PRECAUTIONS:** *Liquefied gas under pressure.* Use piping and equipment adequately designed to withstand pressures to be encountered. *Gas can cause rapid suffocation due to oxygen deficiency.* Store and use with adequate ventilation. Close cylinder valve after each use; keep closed even when empty. Be sure to read and understand all labels and instructions supplied with all containers of this product. *Never work on a pressurized system.* If there is a leak, close the cylinder valve. Blow down the system in an environmentally safe manner in compliance with all federal, state, and local laws, then repair the leak. *Never ground a compressed gas cylinder or allow it to become part of an electrical circuit.*

**NOTE:** Prior to using any plastics, confirm their compatibility with Sulfur Hexafluoride.

**MIXTURES:** When you mix two or more gases or liquefied gases, you can create additional, unexpected hazards. Obtain and evaluate the safety information for each component before you produce the mixture. Consult an industrial hygienist, or other trained person when you evaluate the end product. Remember, gases and liquids have properties that can cause serious injury or death.
HAZARD RATING SYSTEMS:

<table>
<thead>
<tr>
<th>NFPA RATINGS:</th>
<th>HMIS RATINGS:</th>
</tr>
</thead>
<tbody>
<tr>
<td>HEALTH = 1</td>
<td>HEALTH = 0</td>
</tr>
<tr>
<td>FLAMMABILITY = 0</td>
<td>FLAMMABILITY = 0</td>
</tr>
<tr>
<td>REACTIVITY = 0</td>
<td>REACTIVITY = 0</td>
</tr>
<tr>
<td>SPECIAL = None</td>
<td></td>
</tr>
</tbody>
</table>

STANDARD VALVE CONNECTIONS FOR U.S. AND CANADA:

- **THREADED:** CGA-590
- **PIN-INDEXED YOKE:** Not applicable
- **ULTRA-HIGH-INTEGRITY CONNECTION:** CGA-716

Use the proper CGA connections. **DO NOT USE ADAPTERS.**

Additional limited-standard connections may apply. See CGA Pamphlet V-1.

Ask your supplier about free Praxair safety literature as referenced on the label for this product; you may also obtain copies by calling 1-800-PRAXAIR. Further information about Sulfur Hexafluoride can be found in the following pamphlets published by the Compressed Gas Association, Inc. (CGA), 1725 Jefferson Davis Highway, Arlington, VA 22202-4102, Telephone (703) 412-0900.

- **AV-1** Safe Handling and Storage of Compressed Gases
- **P-1** Safe Handling of Compressed Gases in Containers
- **P-14** Accident Prevention in Oxygen-Rich, Oxygen-Deficient Atmospheres
- **SB-2** Oxygen-Deficient Atmospheres
- **V-1** Compressed Gas Cylinder Valve Inlet and Outlet Connections
- **---** Handbook of Compressed Gases, Third Edition

Praxair asks users of this product to study this Material Safety Data Sheet (MSDS) and become aware of product hazards and safety information. To promote safe use of this product, a user should (1) notify employees, agents and contractors of the information on this MSDS and of any other known product hazards and safety information, (2) furnish this information to each purchaser of the product, and (3) ask each purchaser to notify its employees and customers of the product hazards and safety information.
The opinions expressed herein are those of qualified experts within Praxair, Inc. We believe that the information contained herein is current as of the date of this Material Safety Data Sheet. Since the use of this information and the conditions of use of the product are not within the control of Praxair, Inc., it is the user's obligation to determine the conditions of safe use of the product.